

# THE INTRAMOLECULAR DYNAMICS OF A 'RIGID YET TWISTY' FERROCENYL

## TETRAPHOSPHINE – SERVED WITH SOME $^{31}\text{P}$ -NMR DELICACY

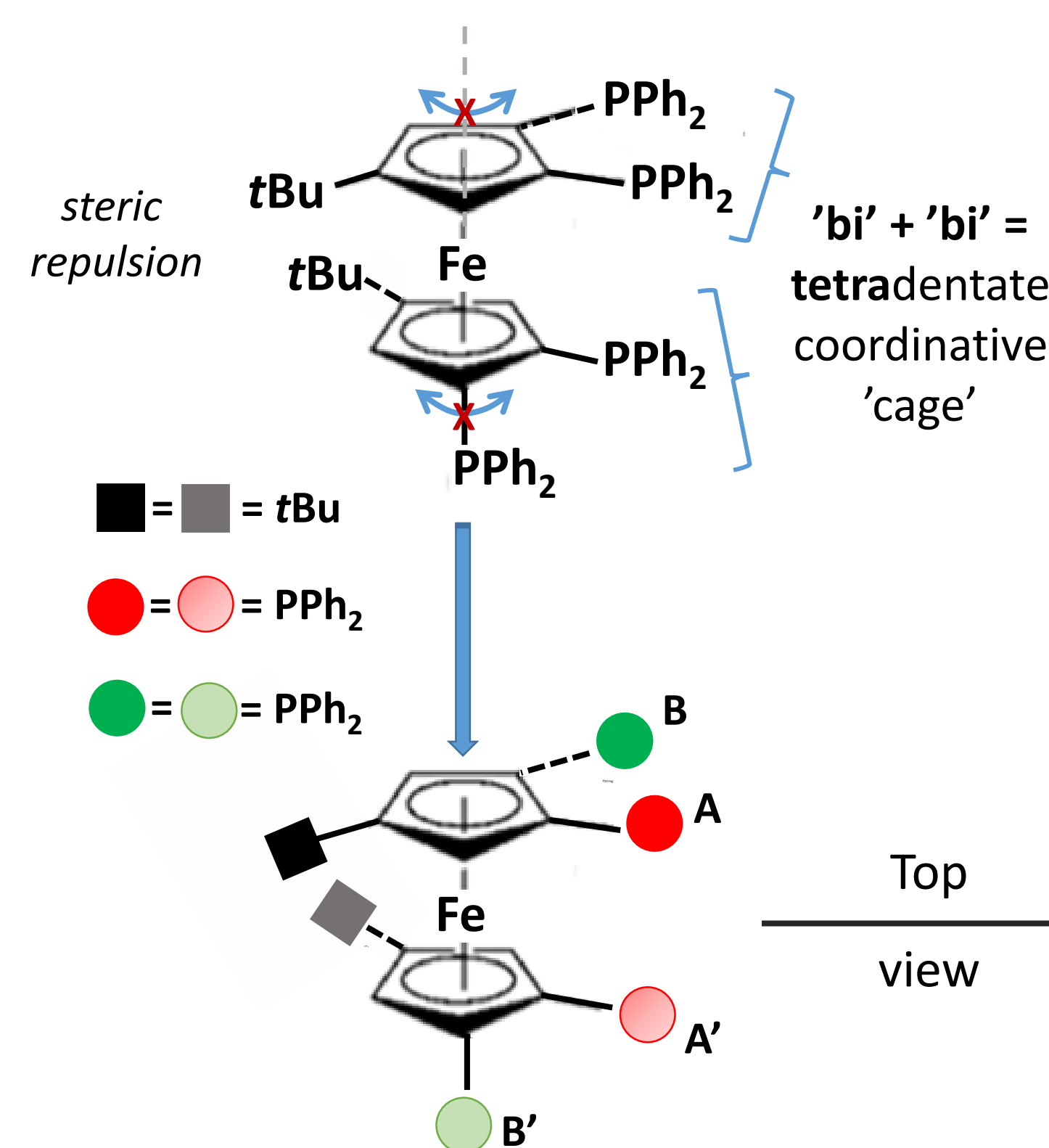
B. Kovács<sup>1</sup>, C. A. Urbina Blanco<sup>2</sup>, M. Saeys<sup>2</sup>, J.-C. Hierso<sup>3</sup>, José C. Martins<sup>1</sup>

<sup>1</sup> NMR AND STRUCTURE ANALYSIS RESEARCH UNIT, DEPARTMENT OF ORGANIC AND MACROMOLECULAR CHEMISTRY, GHEENT UNIVERSITY, GHEENT, BELGIUM <sup>2</sup> LABORATORY FOR CHEMICAL TECHNOLOGY, DEPARTMENT OF MATERIALS, TEXTILES AND CHEMICAL ENGINEERING, GHEENT UNIVERSITY, GHEENT, BELGIUM <sup>3</sup> LABORATOIRE DE SYNTHÈSE ET D'ELECTROSYNTHÈSE ORGANOMÉTALLIQUES, UNIVERSITÉ DE BOURGOGNE, DIJON, FRANCE

### Introduction: multidentate ferrocenyl phosphines

- Multidentate ferrocenyl phosphines offers multiple coordinative sites for a  $\equiv\text{P}$ : lone electron pair acceptor
- The Cp rings are typically equipped with bulky substituents  $\rightarrow$  the antiparallel ring rotation (twisting) is hindered  $\rightarrow$  a permanent coordinative 'cage' is formed consisted of multiple coordinative sites
- Application:** molecular cluster, nanoparticle, surface stabilizers/activators [1], and common ligands in transition metal catalyzed Suzuki cross-coupling reactions [2]

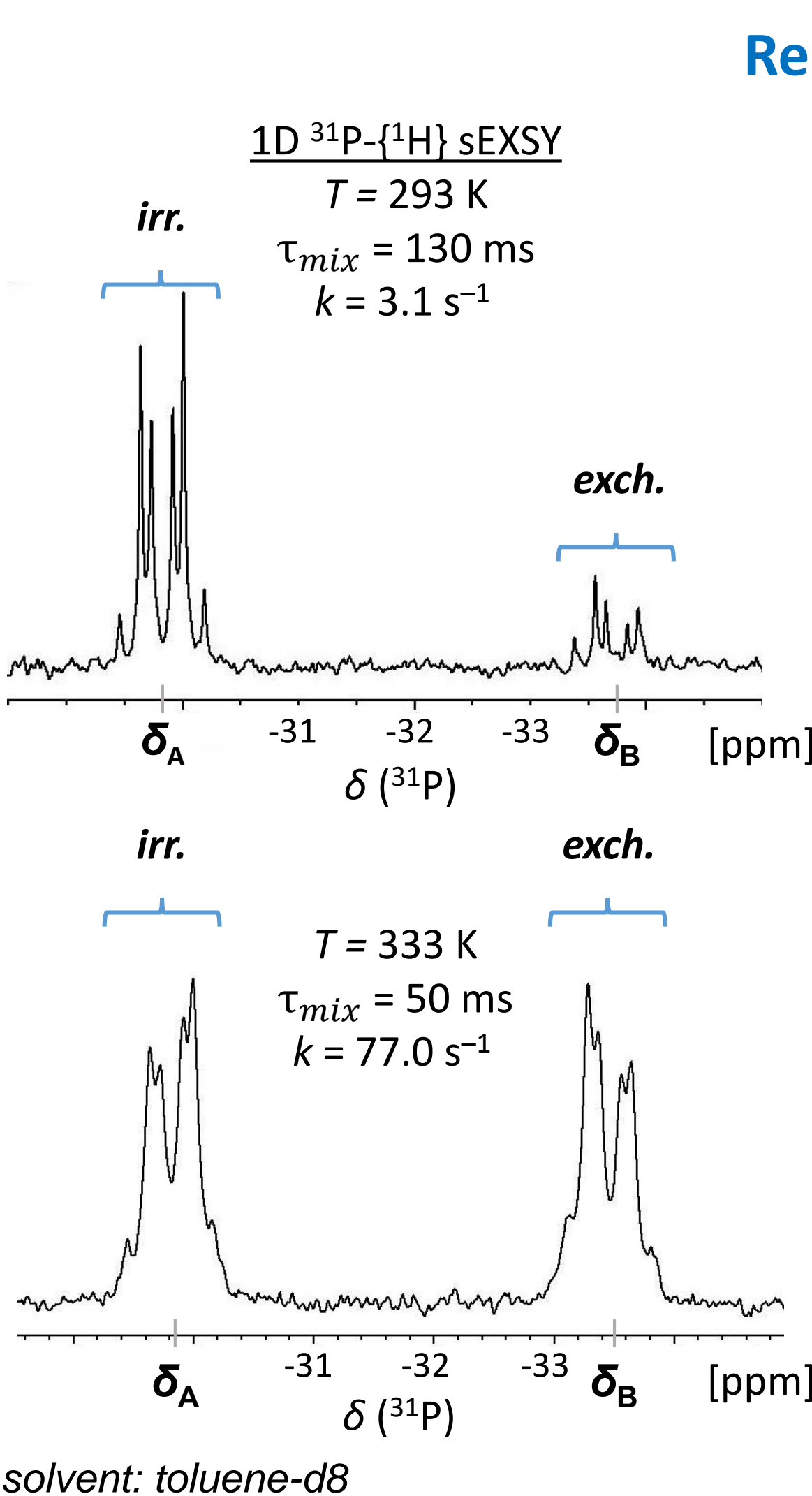
### Investigated molecule: $\text{Fc}(\text{P})_4\text{tBu}$



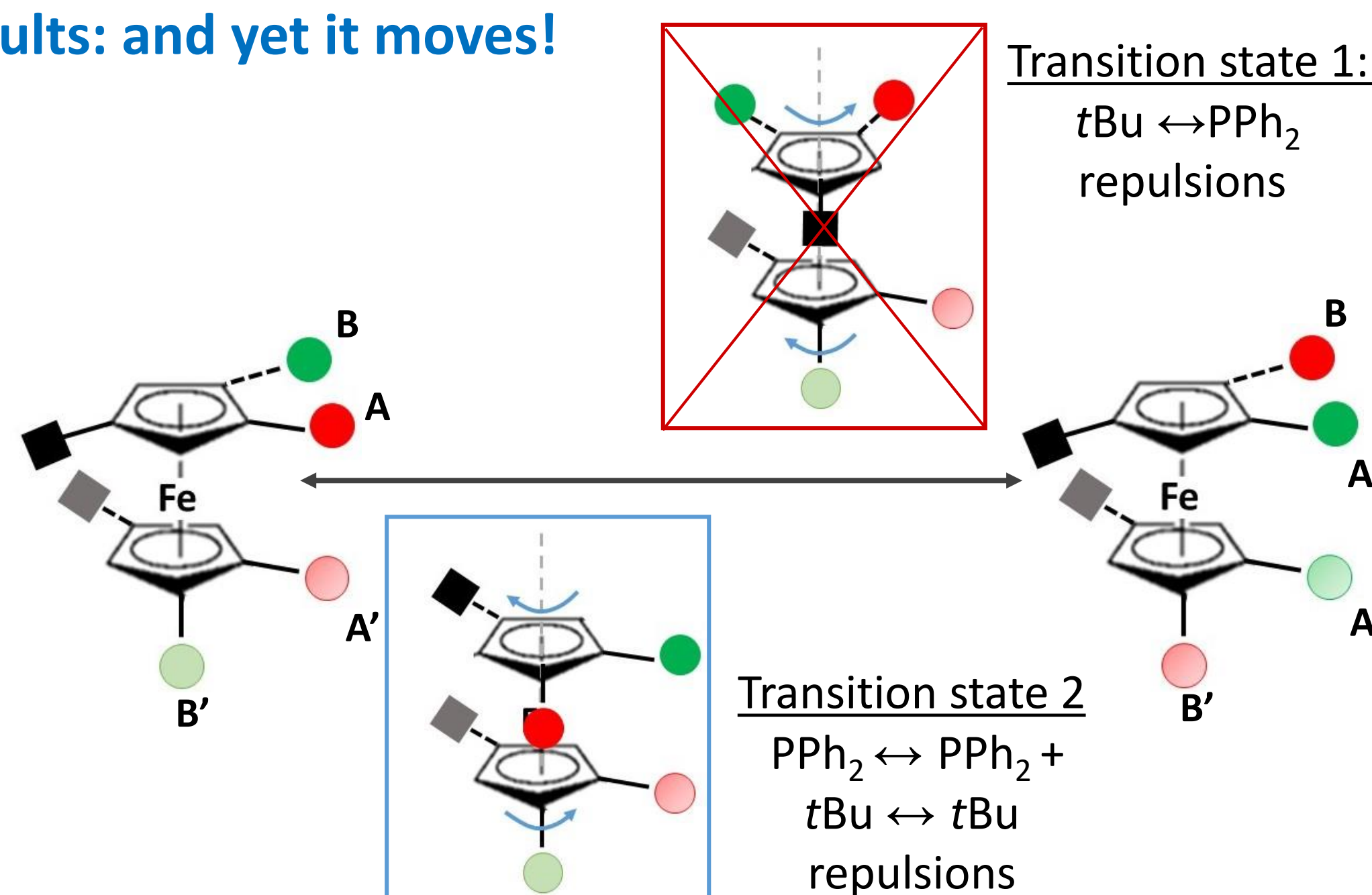
### Objectives: hindered motions to reveal

- Conception of 'static, rigid' conformation (see left) to be refined
- Hindered antiparallel Cp rotations to characterize – both in a qualitative and quantitative fashion
- Sufficiently long time scale is needed ...  $\rightarrow$

### NMR spectroscopy!



### Results: and yet it moves!

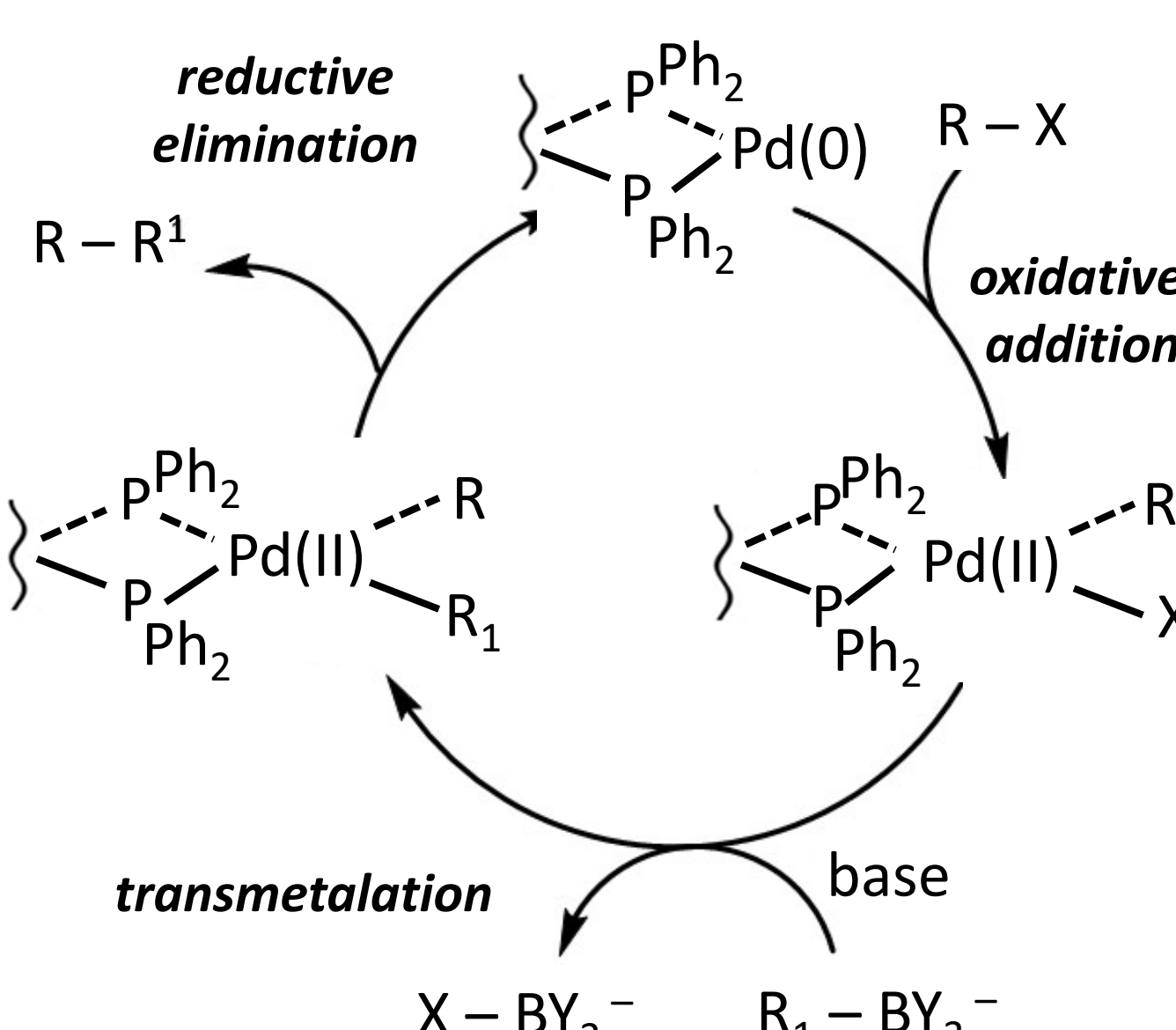
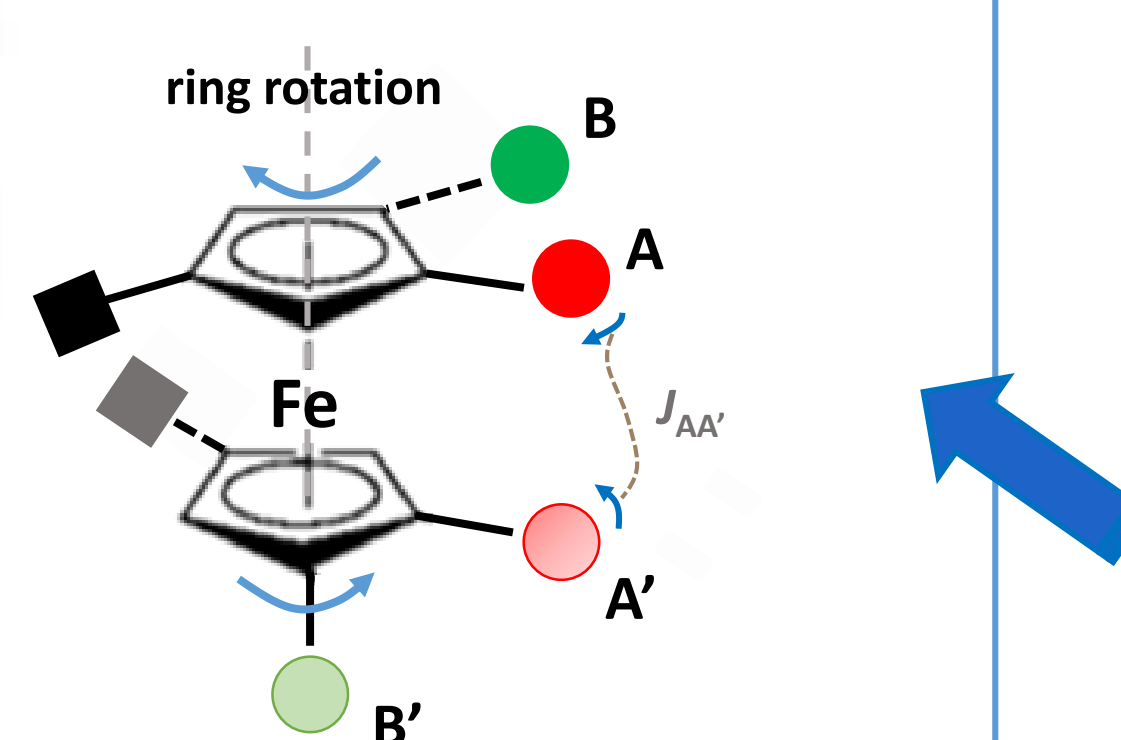


- From  $k_{\text{rot}}(T) \rightarrow \Delta H_{\text{rot}}^\ddagger$  and  $\Delta S_{\text{rot}}^\ddagger$  (via Eyring-plot [3])  
 $\rightarrow \Delta G_{\text{rot}, 298\text{ K}}^\ddagger (\text{exp.}) = 69.0\text{ kcal}\cdot\text{mol}^{-1}$
- According to DFT calculations (PBE1PBE/Def2SVP-w06)  
 $\Delta G_{298\text{ K}}^\ddagger (\text{theor.}) = 68.9\text{ kcal}\cdot\text{mol}^{-1}$  for transition state 2
- $\rightarrow$  Preferred way of (hindered) rotation keeps the coordinative sites close

### $J_{\text{AA}'}$ : through space or not?

$$k_{\text{rot}}(333\text{ K}) = 77.0 \pm 0.2\text{ Hz} > 59.5\text{ Hz} = J_{\text{AA}'}$$

and still AA'BB'!

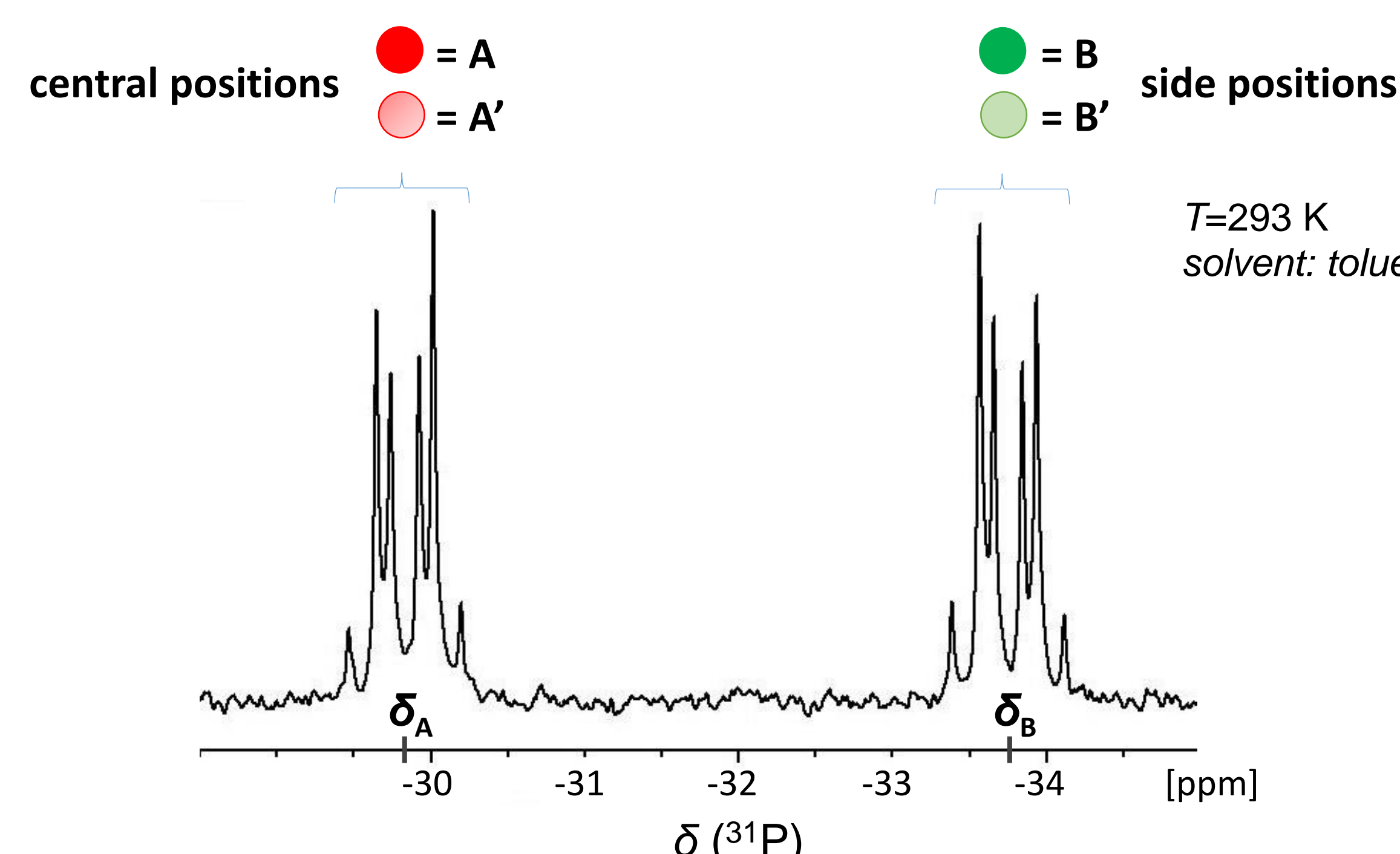


### Reminder on Suzuki cross-couplings

- Metal center is solubilized before oxidative addition by the phosphine
- Bulkiness of the ligand facilitates the reductive elimination step
- Multiple coordinative sites provide longer catalytic lifetime and higher efficiency

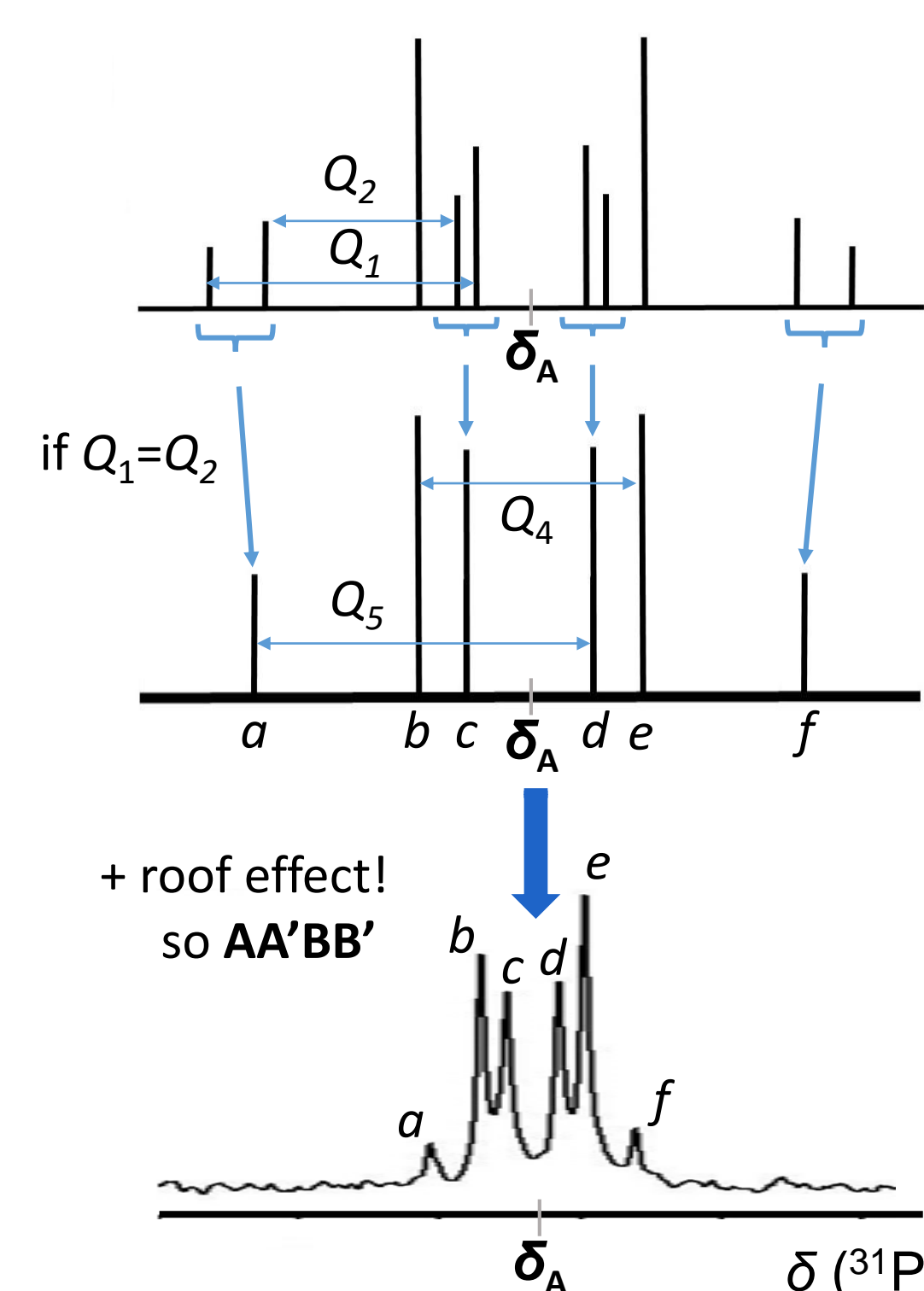
### $^{31}\text{P}$ - $\{^1\text{H}\}$ NMR spectrum of $\text{Fc}(\text{P})_4\text{tBu}^*$

- 2 chemical environments (red and green) for the 4 phosphorus nuclei due to symmetry

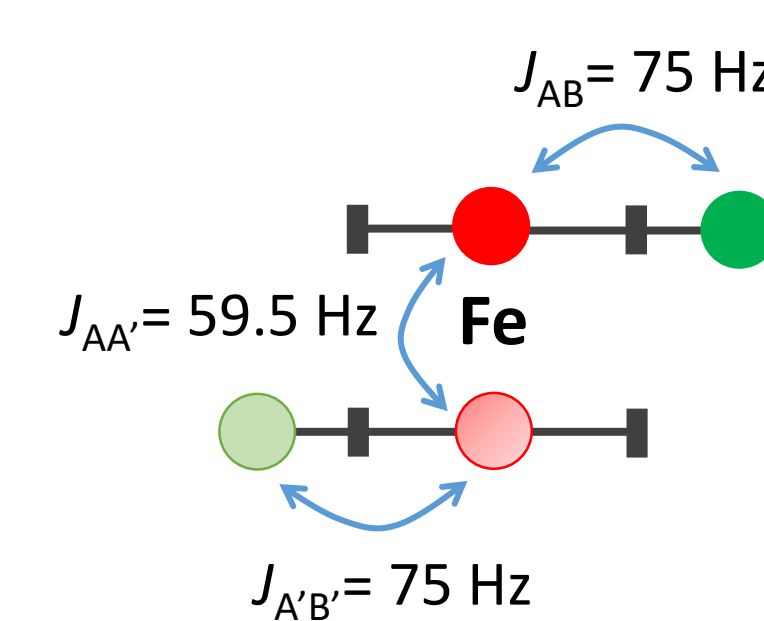


- Both peaks in the  $^{31}\text{P}$ - $\{^1\text{H}\}$  display puzzling patterns  $\rightarrow$  couplings between  $^{31}\text{P}$ -s on different Cp rings exist  $\rightarrow$  4-membered spin system (but  $\text{A}_2\text{X}_2$ ,  $\text{A}_2\text{B}_2$ ,  $\text{AA'XX'}$ , or  $\text{AA'BB'}$ ?)  
 $\rightarrow \text{AA'XX'}$  spin system with subtle second order effects! Centrosymmetric spectrum
- In principle, for each spin 10–10 (spin-allowed) transition frequencies are conceivable if  $J_{\text{AA}'} \neq J_{\text{XX}'} \neq J_{\text{AX}'} = J_{\text{A'X}} \neq J_{\text{AX}} = J_{\text{A'X'}} \neq 0$
- How to read the scalar coupling constant values?

At  $\delta_{\text{A}}$  for instance:



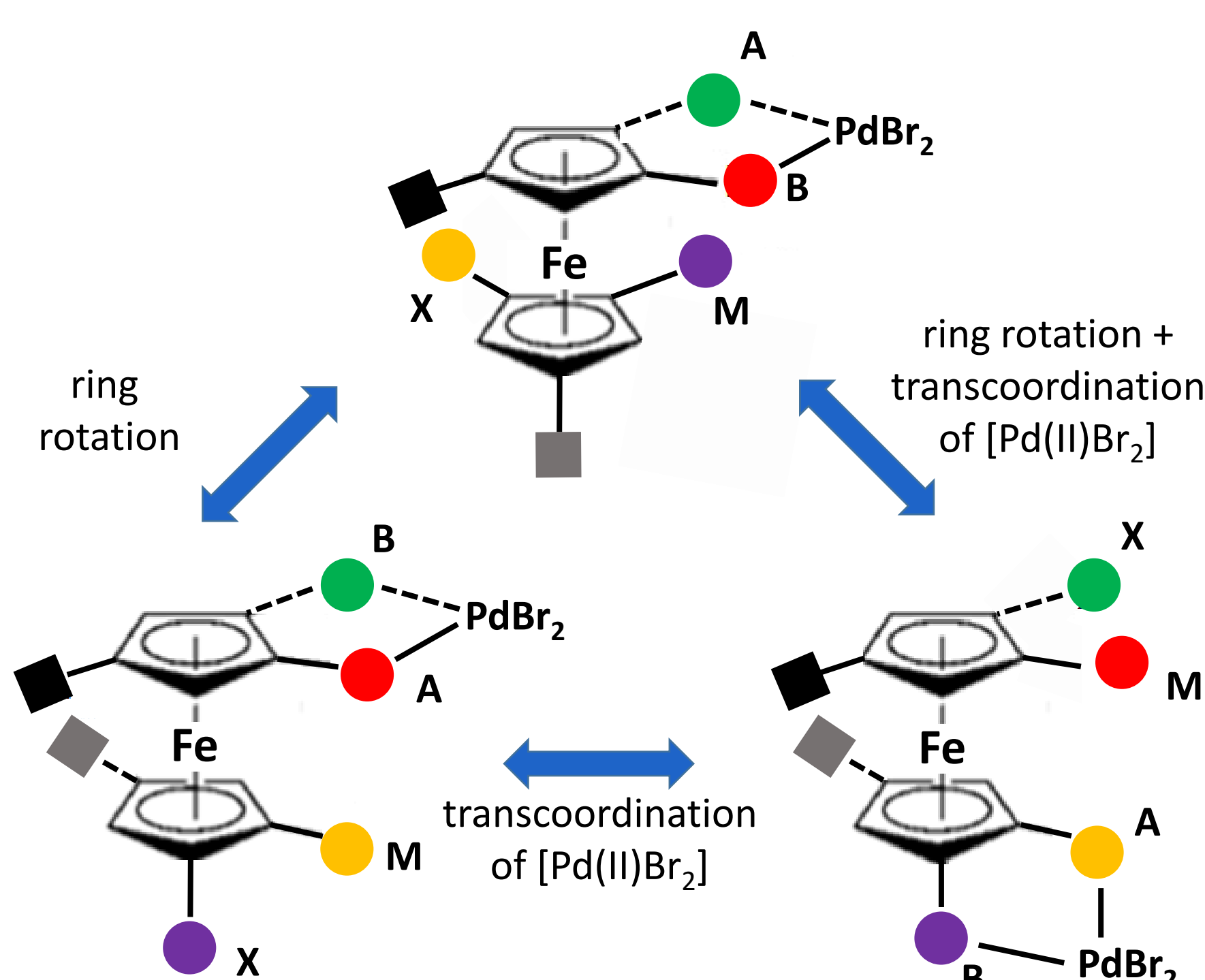
$$\begin{aligned} J_{\text{AX}'} &= J_{\text{A'X}} = \frac{1}{2}(Q_4 - Q_3) = 0\text{ Hz} \\ J_{\text{AA}'} &= \frac{1}{2}(Q_1 + Q_2) = 59.5\text{ Hz} \\ J_{\text{AX}} &= J_{\text{A'X'}} = \frac{1}{2}(Q_3 + Q_4) = 75\text{ Hz} \\ J_{\text{XX}'} &= \frac{1}{2}(Q_1 - Q_2) = 0\text{ Hz} \end{aligned}$$



- $^3J_{\text{AA}'}$  is reported to be through space [4]

### Results II – $[\text{Pd}(\text{II})\text{Br}_2\text{-Fc}(\text{P})_4\text{tBu}]$

- $[\text{Pd}(\text{II})\text{Br}_2\text{-Fc}(\text{P})_4\text{tBu}]$ : 16  $e^-$  state of Pd
- ABMX  $^{31}\text{P}$  spin system (4 peaks)
- 1D  $^{31}\text{P}$  and 2D  $^1\text{H}$  EXSY-s
- $\rightarrow \text{A} \leftrightarrow \text{B} \leftrightarrow \text{M} \leftrightarrow \text{X}$  are in dynamic exchange
- $\rightarrow$  Ring rotation + transcoordination of  $=\text{PdBr}_2$
- Benefits of multiple coordinative sites
- $k(\text{rot}) > k(\text{transc.})$



### Conclusions

- The 'rigidity' of the Cp scaffold in  $\text{Fc}(\text{P})_4\text{tBu}$  was quantified experimentally and *in silico*
- The preferred pathway of the hindered Cp ring twisting supports the existence of a permanent multidentate coordinative 'cage'
- Transition metal transcoordination between distinct coordinative sites was directly shown in case of  $[\text{Pd}(\text{II})\text{Br}_2\text{-Fc}(\text{P})_4\text{tBu}]$

### Acknowledgements

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### References

- [1] Zhang, S., Leem, G. et. al., *J. Am. Chem. Soc.*, 2008, **130**, 113-120 [2] Hierso, J.-C., Fihri, A. et. al., *Organometallics*, 2003, **22**, 4490-4499 [3] Bain, A., Chemical Exchange in NMR, *Prog. Nucl. Magn. Reson. Spectrosc.*, 2003, **43**, 63-103 [4] Hierso, J.-C., Fihri, A., Ivanov, V. V., Hanquet, B., Pirio, N., et. al., *J. Am. Chem. Soc.*, 2004, **126**, 11077-11087

\*Note: the assignments of the  $^{31}\text{P}$  NMR spectra were adapted from [4]. However, the interpretation of the peak multiplicities and the investigation of dynamic properties are novel results and the product of the authors of this work.